



## Original Research Article

# Ground water quality and its health impact analysis in an industrial area

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## A B S T R A C T

### Keywords

Ground water quality, Maraimalar Nagar, Tamilnadu, Industrial runoff,

Ground water quality, especially in areas that immediately surround industrial zones are of increasing interest owing to the nearing proximity of residential zones. This paper is an important report on the ground water quality status near the industrial area of Maraimalai Nagar, Kanchipuram District, Tamil Nadu, India. Industries located in industrial area of Maraimalai Nagar dispose their effluents on land and as a result the nearby shallow open wells get polluted and also the salt content of soil has started building up slowly. Forty representative samples were collected from various sources, such as borewells and open-wells during different seasons over three seasons. However, if the pollution continues unabated it could pose serious problems in the future for the expanding residential zones.

## Introduction

The quality of ground water is of great importance in determining the suitability of particular ground water for a certain use (public water supply, irrigation, industrial applications, power generation etc.). This depends on a large number of individual hydrological, physical, chemical and biological factors. that have acted on the water from the moment it condensed in the atmosphere to the time it is discharged by a well<sup>1,2,3</sup>. The quality of ground water varies with location, depth of water table, season and by the extent and composition of dissolved solids. Generally, higher proportions of dissolved constituents are found in ground water than in surface water because of greater interaction of ground water with various materials in geologic strata. Adverse effects on ground water

quality are a result of anthropogenic activity at ground surface such as unintentionally by agriculture, uncontrolled release of domestic and industrial effluents, by sub-surface or surface disposal of sewage and industrial wastes. It is important to know the geochemistry of the chemical-soil-groundwater interactions in order to assess the fate and impact of pollutant discharged on to the ground<sup>4,5,6,7</sup>. Ground water pollution is intrinsically difficult to detect, since the problem may well be concealed below the surface and monitoring is costly, time consuming and hard to resolve. Pollutants move through several different hydrologic zones as they migrate through the soil to the water table. The contamination of ground water by heavy metals and pesticides has assumed much

significance during recent years due to their toxicity and accumulative behaviour. Industrial disposal of chemicals by surface and sub surface runoff, direct release into natural water bodies or waste, dumped near the factories can be subjected to reaction with percolating rain water and reach the ground water level. Solid wastes are dumped into the ravines of river, as fill in for village ponds and as spread over the grazing or waste lands. The percolating water picks up a large amount of dissolved constituents and reaches the aquifer system and contaminates the ground water<sup>8,9</sup>.

This is a report on the ground water quality status near the industrial area of Maraimalai Nagar, Kanchipuram District, Tamil Nadu. The main objective of this paper is to examine pollution threats, especially to the groundwater resources, around industrial area of Maraimalai Nagar in Kanchipuram District of Taminadu State, India and suggest remedial measures that may also be relevant to other industrial areas. Forty representative samples were collected from various sources, such as borewells and open-wells during different seasons (summer, winter and Post-monsoon) for physico-chemical and bacteriological analysis. Industries located in industrial area of Maraimalai Nagar dispose their effluents on land and as a result the nearby shallow open wells get polluted and also the salt content of soil has started building up slowly.

## **Materials and Methods**

Maraimalai Nagar is the Head quarters of Maraimalai Nagar taluk in Kancheepuram district. Maraimalai Nagar is located in the south direction at a distance of 81 km (12°41'30" latitude and 74°58'00" longitude and 28m elevated from M.S.L), from the city of Chennai. The Municipal Area extends to 58.08 sq.km. Representative

ground water samples were collected from 40 locations at a 10 km radius starting from Maraimalai nagar's Industrial area for the analysis of Physico-chemical and biological parameters.

Drinking water wells, Wells closer to polluting sources like industries, urban wastewater drains, garbage, dumpsites etc. and Wells suspected for natural contaminants like fluoride, iron, arsenic or such pollutants were used as a basic selection criterion for the sampling. Open dug wells, which are not in use or have been abandoned, were not used for sampling. Open dug wells (in use for domestic or irrigation water supply)<sup>10,11</sup>, Tube wells (fitted with a hand-pump or a power-driven pump for domestic water supply or irrigation) and Hand Pumps (used for drinking) were the types of wells from which samples were collected from the sampling locations. The sampling locations are listed in Table 1.

Samples were collected once every week across three seasons namely Summer, Winter and Post-Monsoon respectively. For collection of samples a weighted sample bottle or sampler was used to collect sample from an open well.

Samples from the production tube were collected after running the well for about 5 minutes. For bacteriological samples, when collected from tube wells / hand pump, the spout/outlet of the source was sterilized under flame by spirit lamp before collection of sample in the container<sup>12,13</sup>. From open wells the samples were collected directly in to the pre-sterilized glass bottles. The samples were transported to the laboratory for analysis. The parameters tested, samples collected with sample size and the methods of storage/preservation have been tabulated in Table 2.

## Results and Discussion

The Overall minimum and maximum values for the eight water quality parameters studied are given in Table 3. In addition, the Correlation coefficients for the different

parameters in relation to each other were determined for all three seasons. From the analysis of correlation coefficients, 101 positively correlated values and 46 negatively correlated values were found.

**Table.1** Details of Location with name, type of well and code

S.No	Type of Well	Location Name	Location Code
1	Borewell	Periyar Park	W1
2		Near Ford	W2
3.		Maraimalai Nagar II	W3
4.		Maraimalai Nagar A-II	W4
5.		Peramanur	W5
6.		Selva Nagar Ex-V, Oorapakkam	W6
7		Keezhkaranai	W8
8		GST Road (Hotel),Oorapakkam	W9
9		Oorapakkam (Near Chocklate industries)	W11
10		Opp. Guduvancherry Rly station	W13
11		Near Mahendra World city	W15
12		Guduvancherry	W17
13		Vadamelpakkam	W22
14		Near Engineering College	W26
15		SP Koil	W27
16		Near Mahendra World city	W30
17		Kolathur	W31
18		VOC street, Maraimalai nagar	W32
19		Near Industrial Estate	W33
20		Chengalpattu	W34
21		Guruvanmedu	W38
22		Maraimalai Nagar	W40
1	Handpump	Near Police Station, Maraimalai nagar	W14
2		Near Maraimalai Nagar Municipality	W18
3		Paranur	W19
4		Thirukatchur	W20
5		Villiyampakkam	W21
6		Kattankulathur	W23
7		Peramanur	W24
8		Industrial Estate	W25
9		SP Koil	W28
10		Potheri village	W29
11		Chengalpattu	W35
12		Senneri	W36
13		Karambakkam	W37
14		Ezhichur	W39
1	Openwell	Amman Temple, Oorapakkam	W7
2		Oorapakkam east	W10
3		Oorapakkam	W16
1	Drinking Water	Oorapakkam	W12

**Table.2** Details of sample collection and parameters tested

Parameter	Sampling Collection	Sample Size	Storage/ Preservation
pH	Grab sampling in Plastic /glass container	50 ml	
Electrical Conductivity		50 ml	
Total suspended solids		100 ml	Refrigeration (7 days)
Total Dissolved Solids		100 ml	
BOD		500 ml	Refrigeration (48 hrs)
Hardness		100 ml	Add HNO <sub>3</sub> to pH<2, refrigeration (6 months)
Chlorides		50 ml	Not required (28 days)
Sulphates		100 ml	Refrigeration (28 days)
Sodium, Potassium		100 ml	Not required (6 months)
Nitrates		100 ml	Refrigeration (48 hrs)
Fluorides		100 ml	Not required (28 days)
Alkalinity		100 ml	Refrigeration (14 days)
Ammonia		100 ml	Add H <sub>2</sub> SO <sub>4</sub> to pH>2, refrigeration (28 days)
Hexavalent Chromium, Cr+6	Plastic/ Glass rinse with 1+1 HNO <sub>3</sub>	100 ml	Grab sample; refrigeration (24 hrs)
Heavy Metals (Hg, Cd, Cr, Cu, Fe, Zn, Pb etc.)		500 ml	Filter, add HNO <sub>3</sub> to pH>2; Grab sample (6 months)

**Table.3** Minimum and Maximum values observed across three seasons

S.No	Parameter	PostMonsoon			Winter			Summer			LIMITS
		Min	Max	SD	Min	Max	SD	Min	Max	SD	
1	pH	6.58	7.8	0.24	6.58	7.92	0.31	6.58	8.12	0.38	6.5 – 8.5
2	Turbidity(NTU)	4	30	6.39	4	50	9.67	4	40	7.4	10
3	TDS (mg/L)	458	1620	251.67	492	1700	285.72	492	1692	276.9	500
4	Total Hardness(mg/L)	148	512	84.38	154	512	92.01	168	548	98.3	300
5	Chloride(mg/L)	104	492	91.71	114	524	100.65	114	518	103.2	250
6	Sulphate (mg/L)	40	128	24.20	42	154	29.30	42	156	31.5	200
7	Fluoride(mg/L)	0.02	0.14	0.04	0.02	0.5	0.05	0.02	0.14	0.04	1.0-1.5
8	Iron (mg/L)	0.02	1.28	0.28	0.02	1.64	0.33	0.02	1.58	0.31	0.3

**Table.4** Correlation Co-efficients of major parameters studied

Parameters	Season	pH	TDS	Temp	Zinc	Iron	TH	Turbidity
pH	POST MONSOON	1	-0.05736	0.2675	-0.2869	-0.2296	-0.1497	-0.0686
	WINTER	1	0.0302	0.08469	-0.2998	-0.1663	-0.0764	-0.0216
	SUMMER	1	0.1997	0.3163	-0.2534	-0.04026	0.0407	-0.05970
TDS	POST MONSOON	-0.05736	1	-0.12268	0.1858	0.7689	0.8628	0.6630
	WINTER	-0.4653	1	-0.1362	-0.2954	0.6292	0.8662	0.7662
	SUMMER	0.1997	1	0.05941	0.2070	0.1284	0.2381	0.6664
Temp	POST MONSOON	0.2939	-0.1227	1	0.1161	0.00937	0.1730	-0.0591
	WINTER	0.2990	-0.1362	1	0.0847	-0.0049	-0.1386	-0.0846
	SUMMER	0.2990	-0.1362	1	0.0847	-0.0049	-0.1386	-0.0846
Zinc	POST MONSOON	-0.2869	0.2523	0.1161	1	0.2933	0.4439	0.3849
	WINTER	-0.2998	0.2249	0.0118	1	0.9901	-0.4939	-0.0138
	SUMMER	-0.0612	0.2070	0.1883	1	-0.1360	0.6257	0.2060
Iron	POST MONSOON	-0.2044	0.7721	-0.0057	0.2973	1	0.6140	0.4591
	WINTER	-0.0918	0.6663	0.00607	0.1604	1	0.5542	0.7599
	SUMMER	-0.07573	0.6572	0.01996	-0.0256	1	-0.0744	0.5960
TH	POST MONSOON	-0.1724	0.8621	-0.2173	0.39954	0.6279	1	0.6379
	WINTER	-0.0764	0.8343	-0.1776	0.4338	0.6133	1	0.5963
	SUMMER	0.0407	0.1373	0.0581	0.6364	-0.5114	1	-0.0096
Turbidity	POST MONSOON	-0.1486	0.7612	-0.0591	0.3794	0.8014	0.6307	1
	WINTER	-0.0319	0.1195	-0.0849	-0.0183	0.8581	0.5519	1
	SUMMER	-0.0019	0.6664	-0.0234	0.2060	0.5870	0.0421	1

**Table.5** Water Quality Index of Important Parameters

S.No	Parameter	SEASON		
		Post-Monsoon	Winter Season	Summer Season
1	pH	89	86	87
2	Total Solids	21	21	21
3	Temperature	17	17	17
4	Turbidity	79	77	37
Overall WQI		58 %	52 %	36 %

They are tabulated in Table 4. Correlation coefficients for the different parameters were determined for the summer, winter and Post - Monsoon seasons. From the analysis  $r \geq 0.06$  is significant cases. There are 101 number of positive correlated values and 46 number of negatively correlated values as found in the above tables.

Highest correlation was found among (a) In post- monsoon Seasons: pH and Temperature, TDS and Iron, TDS TH, TDS and Turbidity, Iron and TH (b) Winter Season: Turbidity and Iron Between TDS and Iron, TDS and Turbidity, Iron and TDS. TH and Turbidity, Turbidity and Iron and (c) In summer season between TDS and Turbidity , TH and Zinc, Iron and Turbidity, Turbidity and Iron, Turbidity and TDS.

On ascertaining the Water Quality Index, the percentage of the samples were found as “bad category” in Summer, “medium category” in Post-monsoon, and “medium category” in the Winter. The results are tabulated in Table 5. It could be generalized that Water taken in the industrial area of Maraimalai Nagar town within 5 Km radius was found to be un-potable, whereas the water taken from the 10Km radius of the industrial town was found to be potable.

This study found that continuous disposal of industrial effluents on land, which has limited capacity to assimilate the pollution load, has led to groundwater pollution. Ground water quality surrounding the industrial areas has deteriorated, and the application of polluted groundwater for potability has resulted in increased salt content of soils.

In some locations drinking water wells (deep bore wells) also have high concentration of salts. Since the local municipality is supplying drinking water to households the impact in the domestic sector has been minimised. It has also been noticed that in some locations industries are supplying drinking water to the affected households. However, if the pollution continues unabated it could pose serious problems in the future. Hence, our research plays an experimental work on ground water quality assessment in the nearby industrial area of Maraimalai nagar in Kanchipuram district of Tamilnadu.

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